

Calorimeter Accessory

Product Number: ENCALMT



Overview

The **Calorimeter** contains a heat source that can deliver a heat flux, at a distinct temperature, into a sample and a temperature-measuring device that can read the resultant temperature change.

Typical experiments

- Determining the Enthalpy of a Chemical Reaction
- The Enthalpy of Neutralization of Phosphoric Acid

How does it work?

To find the enthalpy change per mole of a substance A in a reaction between two substances A and B, the substances are separately added to a calorimeter and the initial and final temperatures (before the reaction has started and after it has finished) are noted. Multiplying the temperature change by the mass and specific heat capacities of the substances gives a value for the energy given off or absorbed during the reaction. Dividing the energy change by how many moles of A were present gives its enthalpy change of reaction.

Technical support

Please contact the Fourier technical support team as follows:

Web: http://fourieredu.com/support/ Email: support@fourieredu.com

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